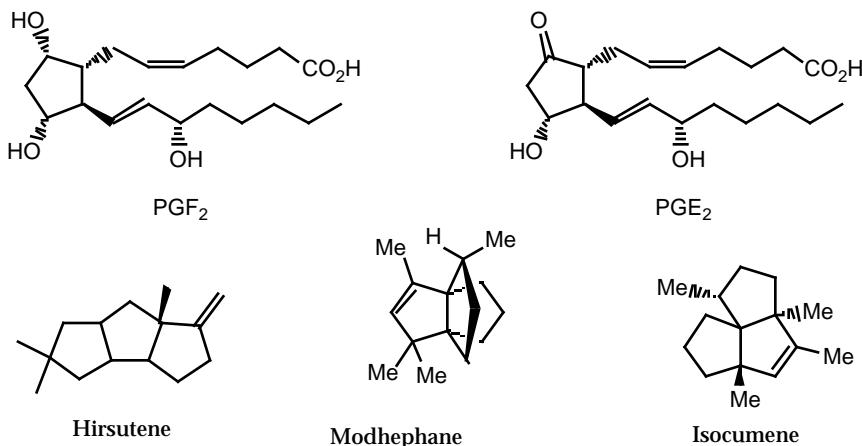


## 5 Membered Rings

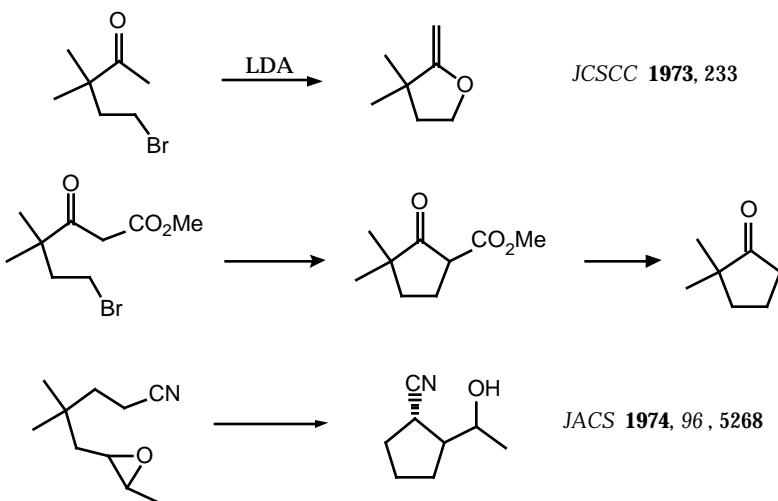


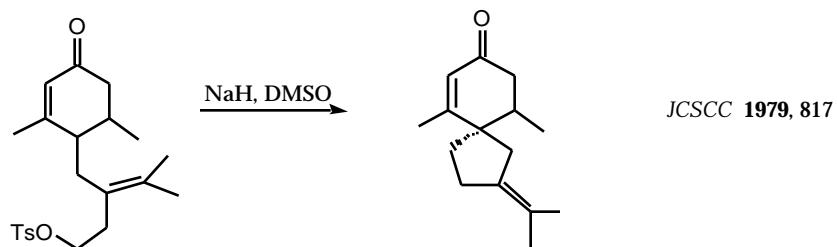
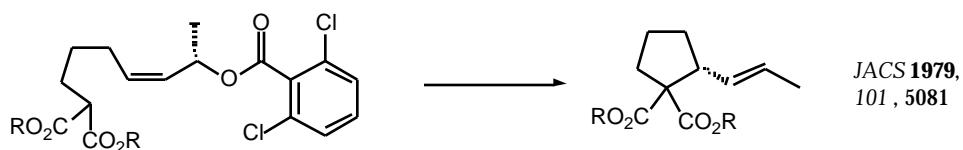
1. Intramolecular S<sub>N</sub>2 Reactions
2. Intramolecular Aldol Condensation and Michael Addition
3. Intramolecular Wittig Olefination
4. Ring Expansion and Contraction Reactions
  - a. 3 5
  - b. 4 5
  - c. 6 5
5. 1,3-Dipolar additions
6. Nazarov Cyclization
7. Arene-Olefin Photocyclization
8. Radical Cyclizations
9. Others

Synthesis **1973**, 397; ACIEE **1982**, 21, 480;

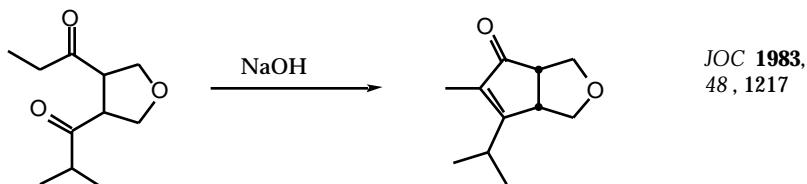
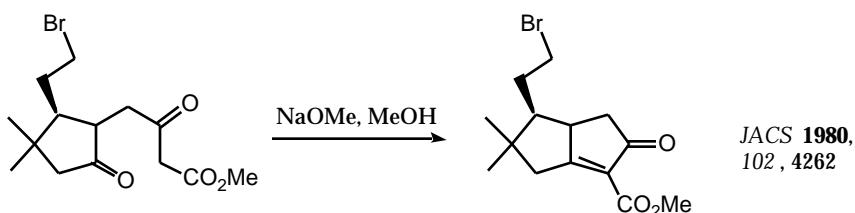
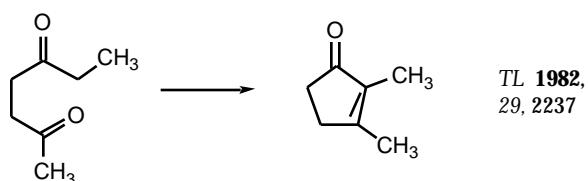
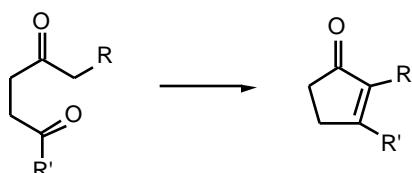
### Intramolecular S<sub>N</sub>2 Reaction

### 5-exo-tet: favored

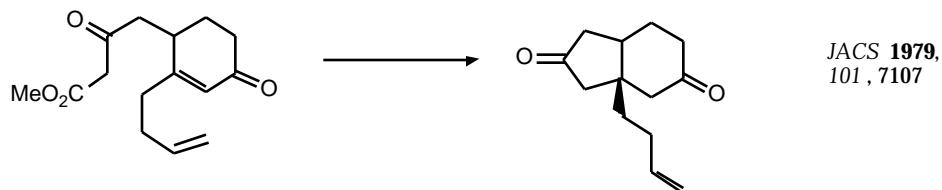




*Intramolecular Aldol Condensation 5-exo-trig: favored intramolecular aldol condensation of 1,4-diketones*

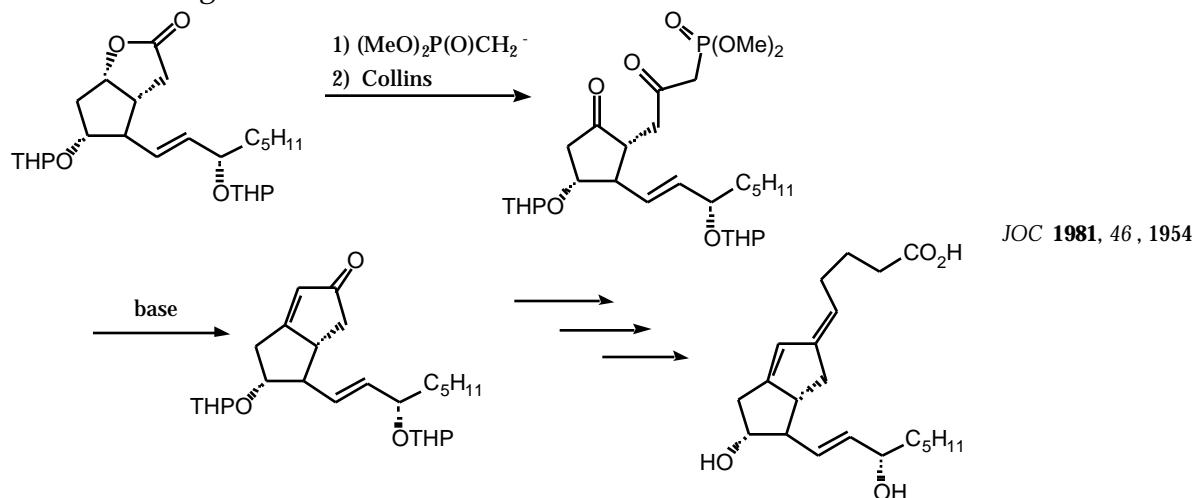


*Intramolecular Michael Addition 5-exo-tet: favored Organic Reactions 1995, 47, 315-552*



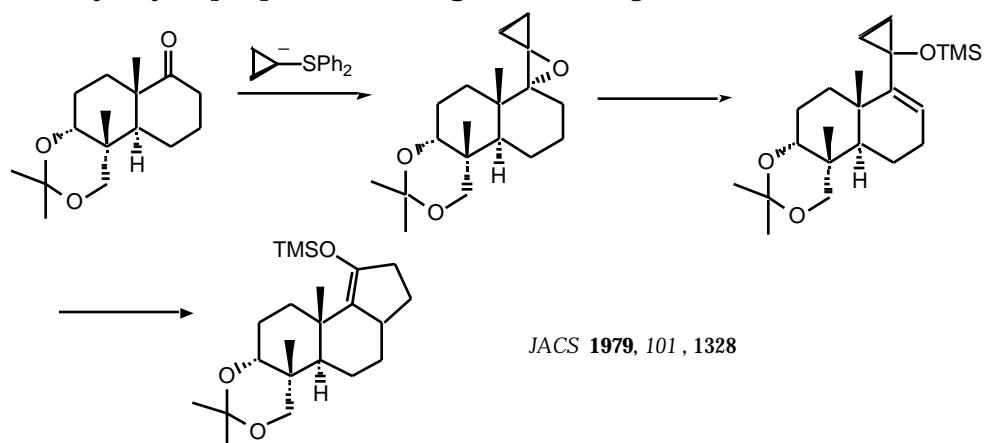
## Intramolecular Wittig Olefination

## Tetrahedron 1980, 36, 1717

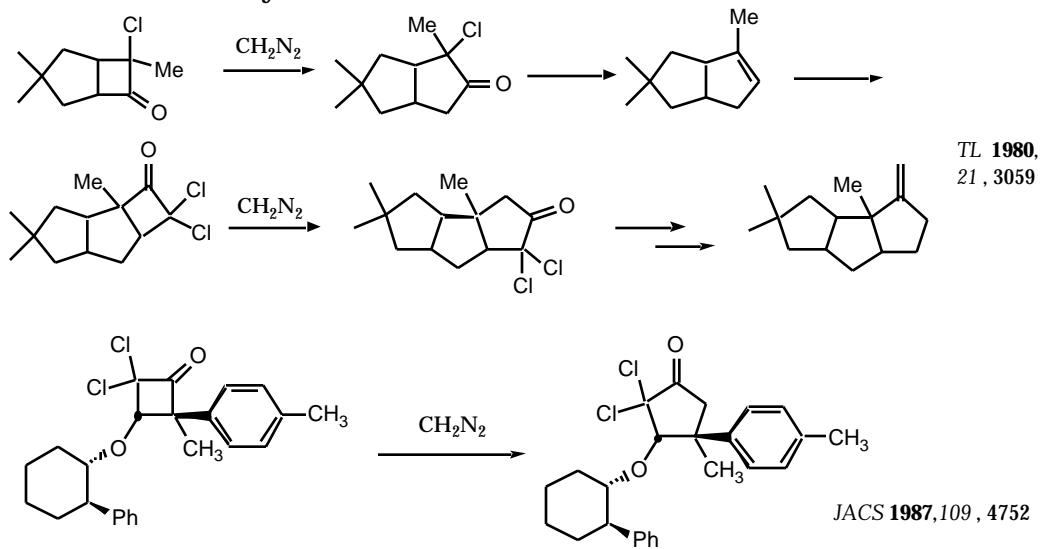


## Ring Expansion Reactions

## - 3 5: Vinyl Cyclopropane Rearrangement      Organic Reactions 1985, 33, 247.

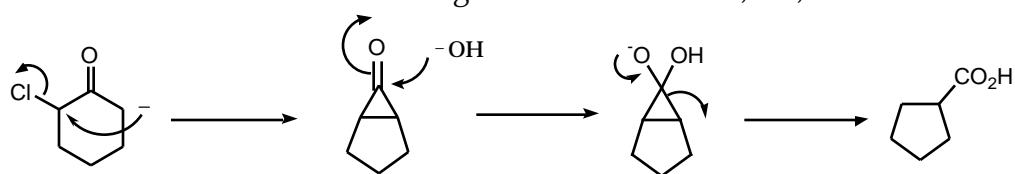


## - 4 5: Reaction of cyclobutanones with Diazomethane

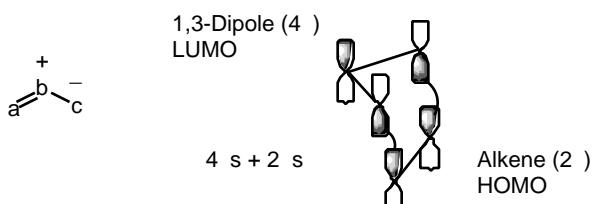


## Ring Contraction Reactions

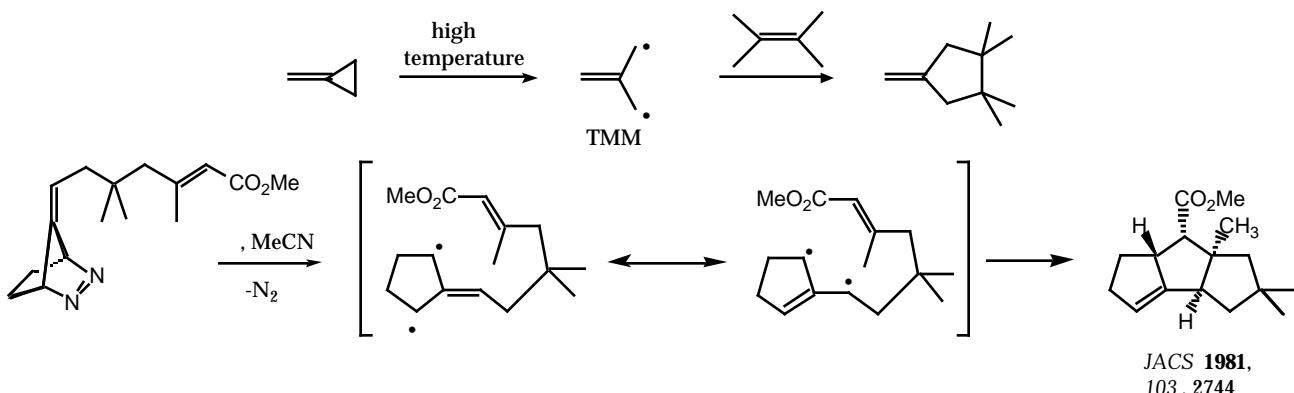
## - 6 5: Favorskii Reaction

Organic Reactions **1960**, 11, 261

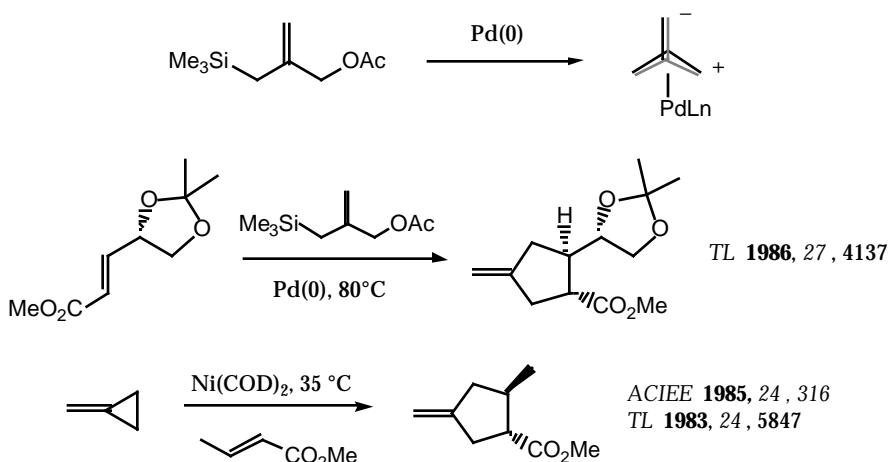
1,3-Dipolar Addition to Olefins

1,3-Dipolar Cycloaddition Chemistry, vol 1 & 2 (A. Padwa ed.) (Wiley, NY 1984); ACIEE **1977**, 16, 10. Chem Rev. **1998**, 98, 863.

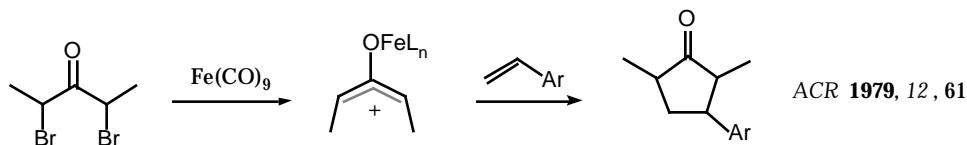
## - trimethylenemethane (TMM)

ACIEE **1986**, 25, 1. Synlett **1992**, 107.

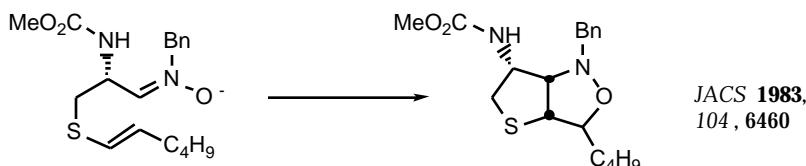
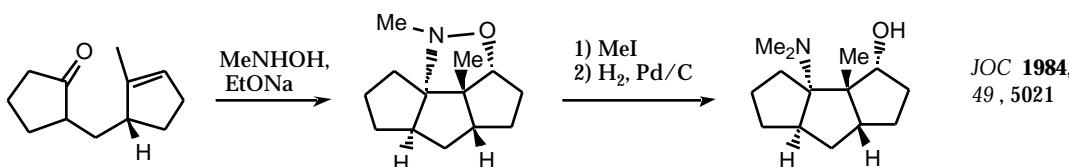
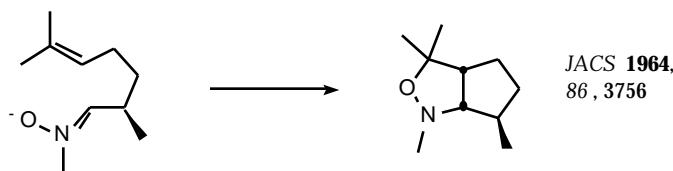
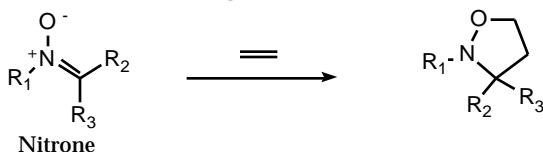
note: TMM usually reacts poorly w/ electron deficient olefins



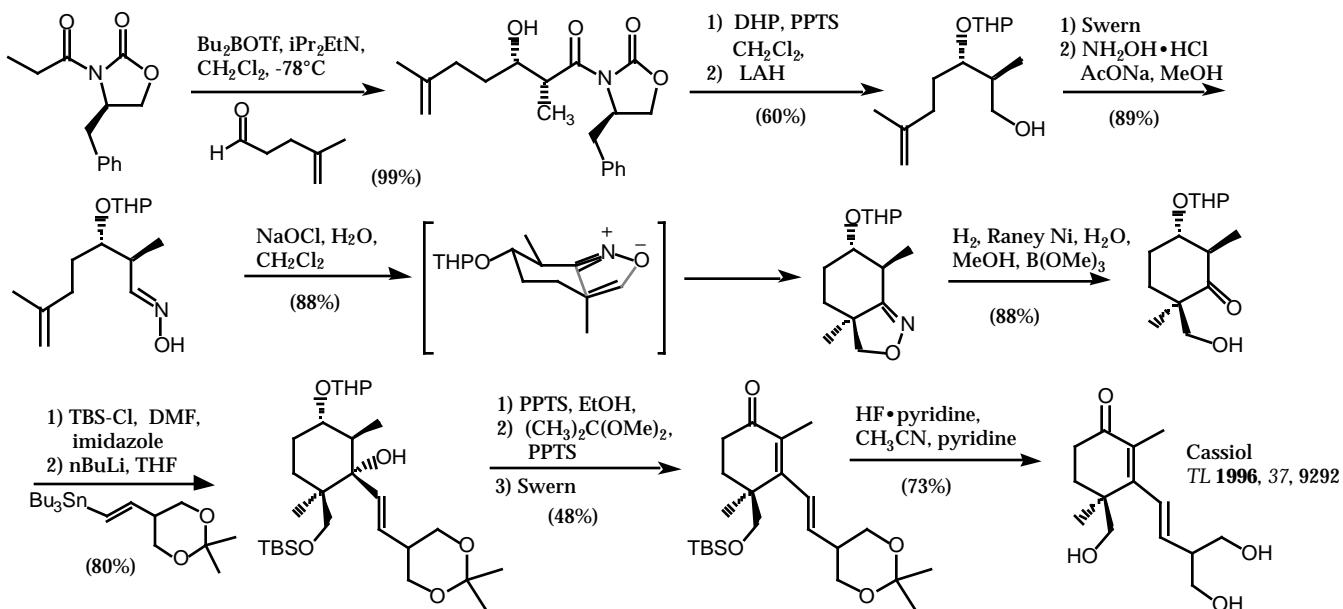
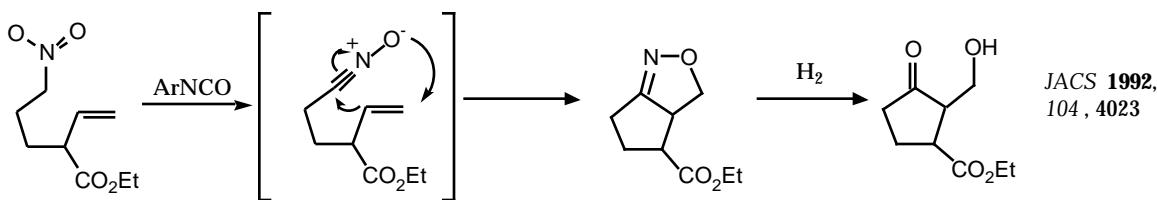
## - , '-dihaloketones

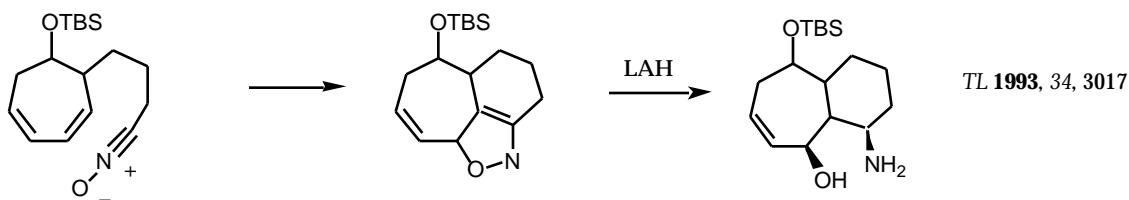


- nitrones    ACR 1979, 12, 396; Organic Reactions, 1988, 36, 1



- nitrile oxides

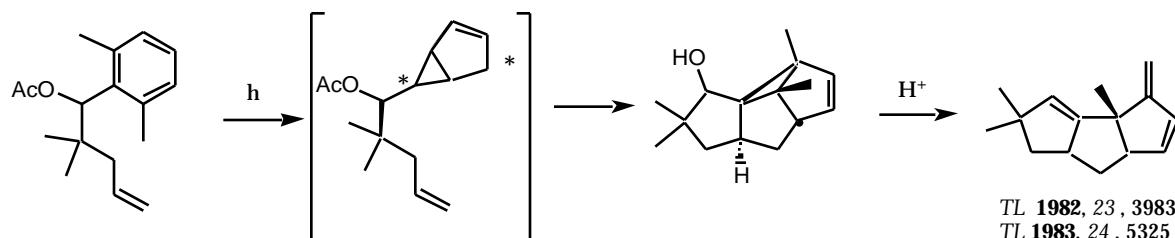
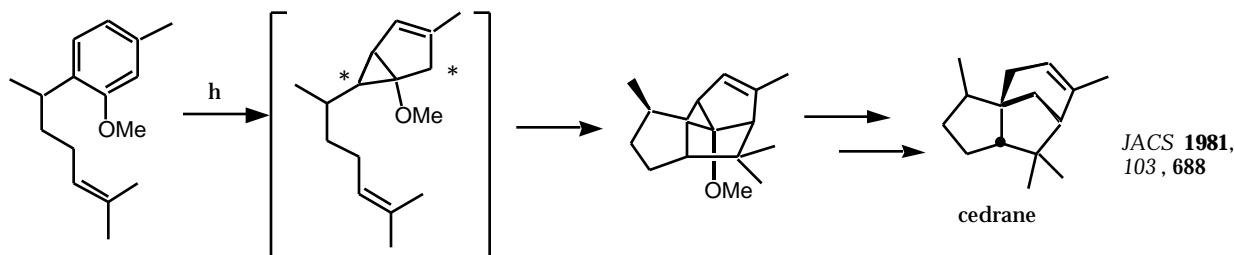
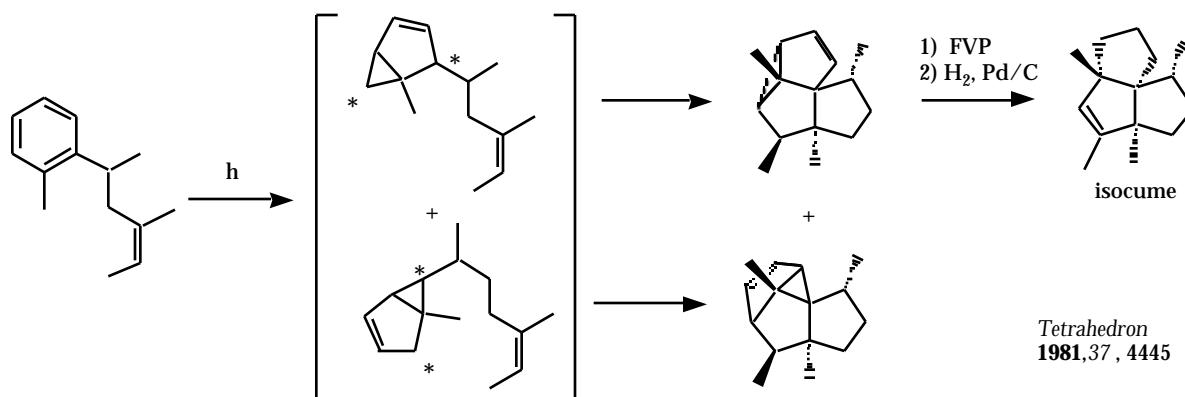
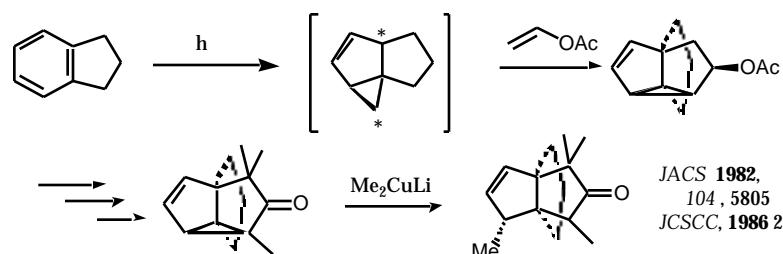
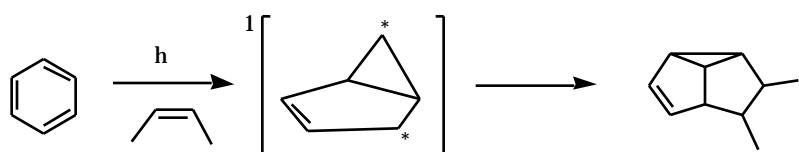




Arene -Olefin Photocyclization

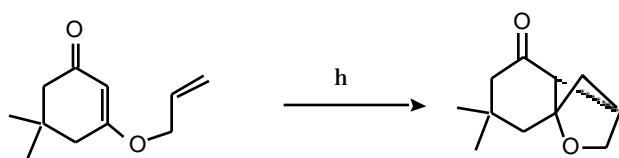
Organic Photochemistry **1989**, *10*, 357

- the photochemistry of benzene is dominated by the singlet state

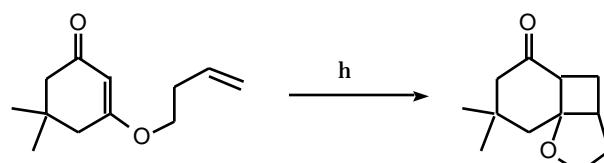


## Intramolecular Photochemical [2+2]

## "Rule of Five"



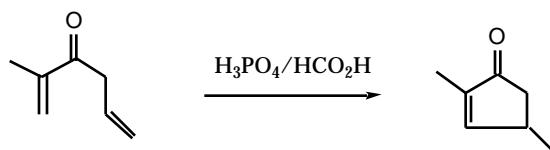
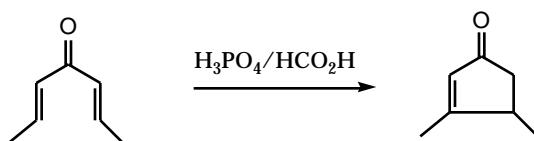
*JOC 1975, 40, 2702  
JOC 1979, 44, 1380*



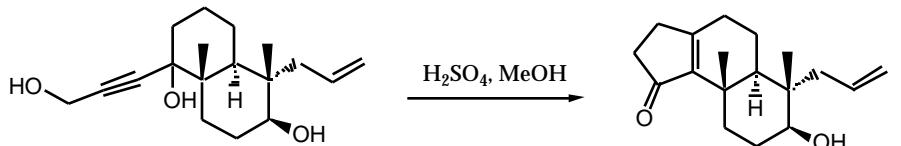
## Nazarov Cyclization

review: *Synthesis* **1983**, 429

## - cyclization of allyl vinyl or divinyl ketones

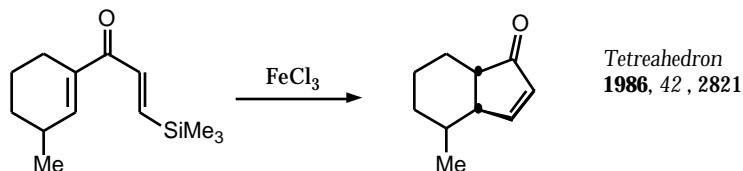


## - 1,4-hydroxy-acetylenes



*JOC 1989,  
54, 3449*

## - Silicon-Directed Nazarov



*Tetrahedron  
1986, 42, 2821*

- Tin -directed Nazarov    *TL 1986, 27, 5947*

## Radical Cyclization

B. Giese *Radicals in Organic Synthesis: Formation of Carbon-Carbon Bonds*

(Pergamon Press; NY) **1986**; *Bull. Soc. Chim. Fr.* **1990**, 127, 675; *Tetrahedron* **1981**, 37,

**3073**; *Tetrahedron* **1987**, 43, 3541; *Advances in Free Radical Chemistry* **1990**, 1, 121.

*Organic Reactions* **1996**, 48, 301-856.

## Radical Addition to multiple bonds:

1. Free radical addition is a two stage process involving an addition step followed by an atom transfer step.
2. In general, the preferred regioselectivity of the addition is in a manner to give the most stable radical (thermodynamic control)

### Advantages of free radical reactions:

1. non-polar, little or no solvent effect
2. highly reactive- good for hindered or strained systems
3. insensitive to acidic protons in the substrates (i.e. hydroxyl groups do not necessarily need to be protected)

### Mechanism of radical chain reactions

1. initiation
2. propagation
3. termination (bad)

### Formation of carbon centered radicals:

tin hydride reduction of

alkyl, vinyl and aryl halides,

alcohol derivatives:

xanthates, thionocarbonate, thiocarbonylimidazolides

organoselenium & boron compounds

carboxylic acid derivatives (Barton esters)

reduction of organomercurials

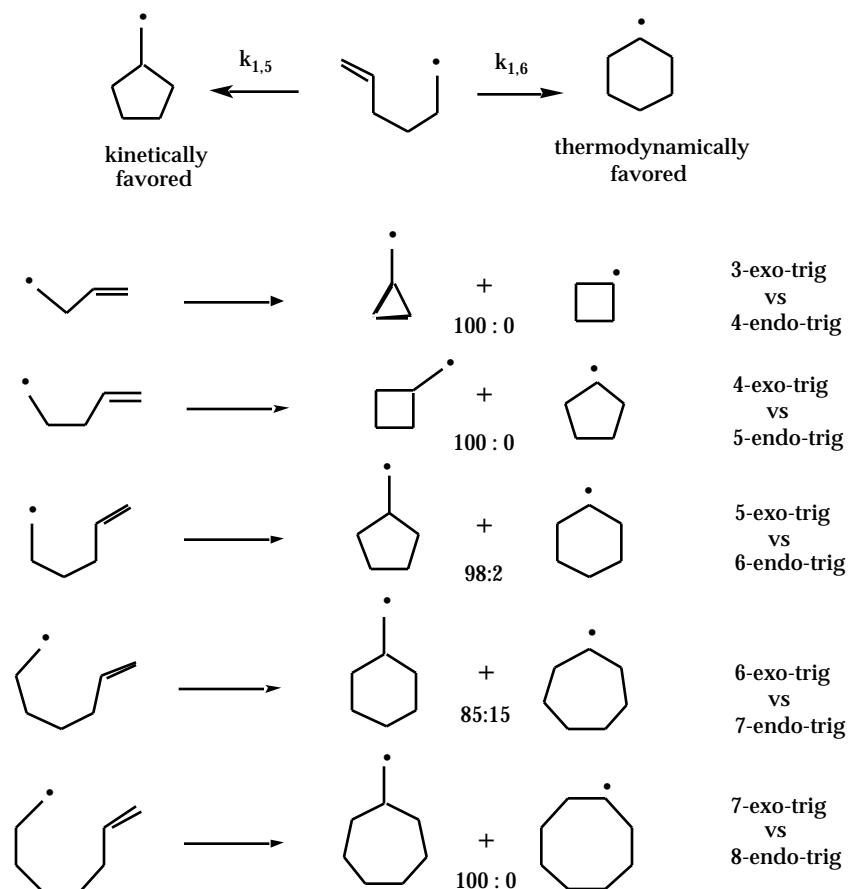
thermolysis of organolead compounds

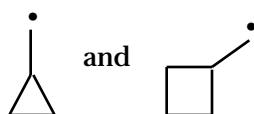
thermolysis or photolysis of azoalkanes.

### Radical Ring Closure

For irreversible ring closure reaction, the kinetic product will predominate.

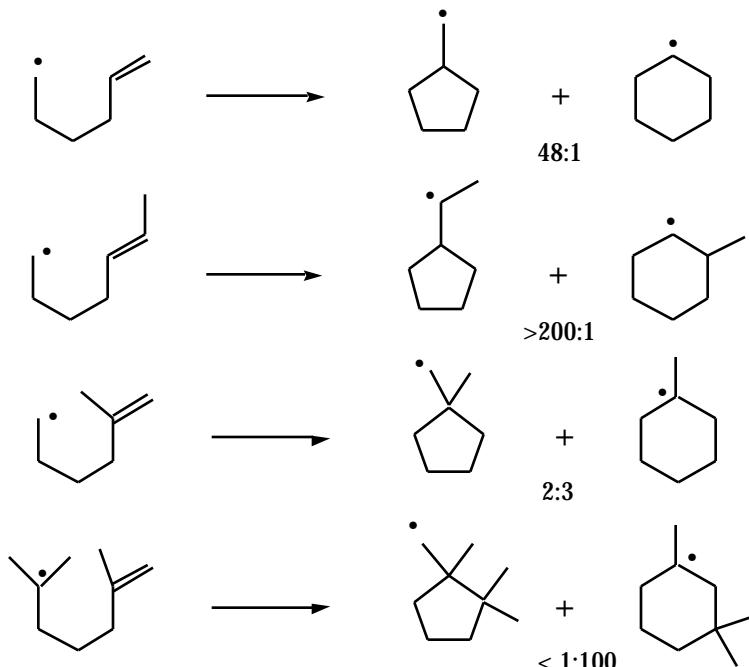
Both the 5-exo-trig and 6-endo trip are favored reactions, with the 6 exo-trig mode producing the most stable radical. However, the 5-exo-trig is about 50 time faster





radicals open up fast and are not synthetically useful;  
often used as probes for radical reaction

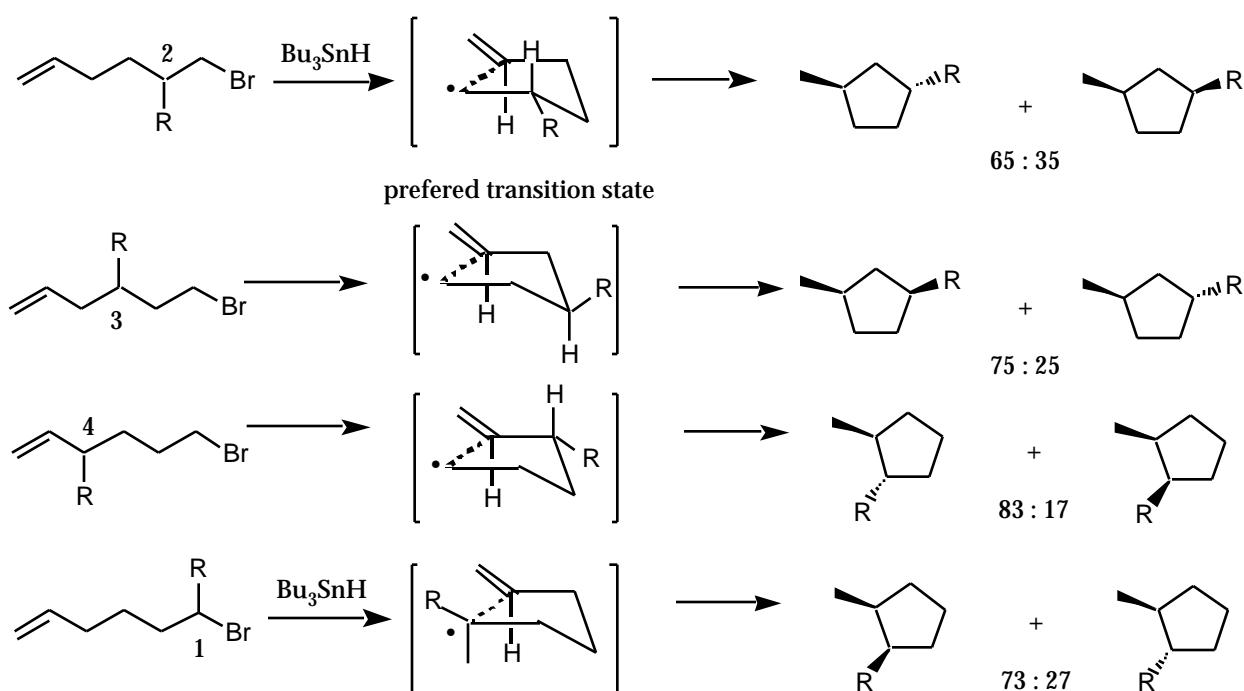
### Effects of substituent on the regiochemistry of the 5-hexenyl radical cyclization

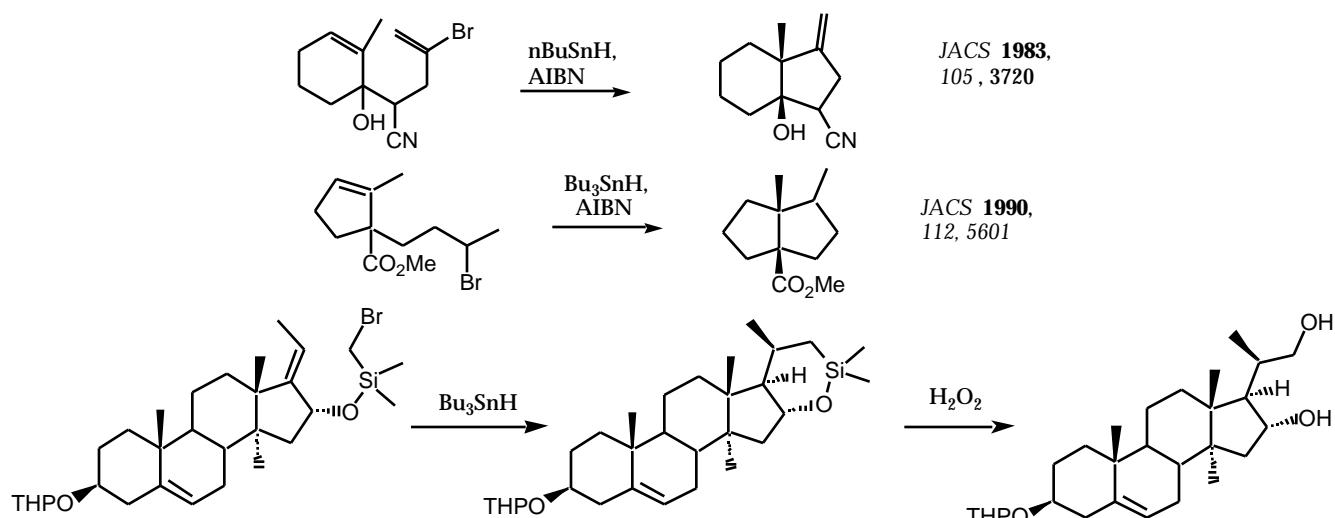


### Stereochemistry of 5-hexenyl radical cyclization

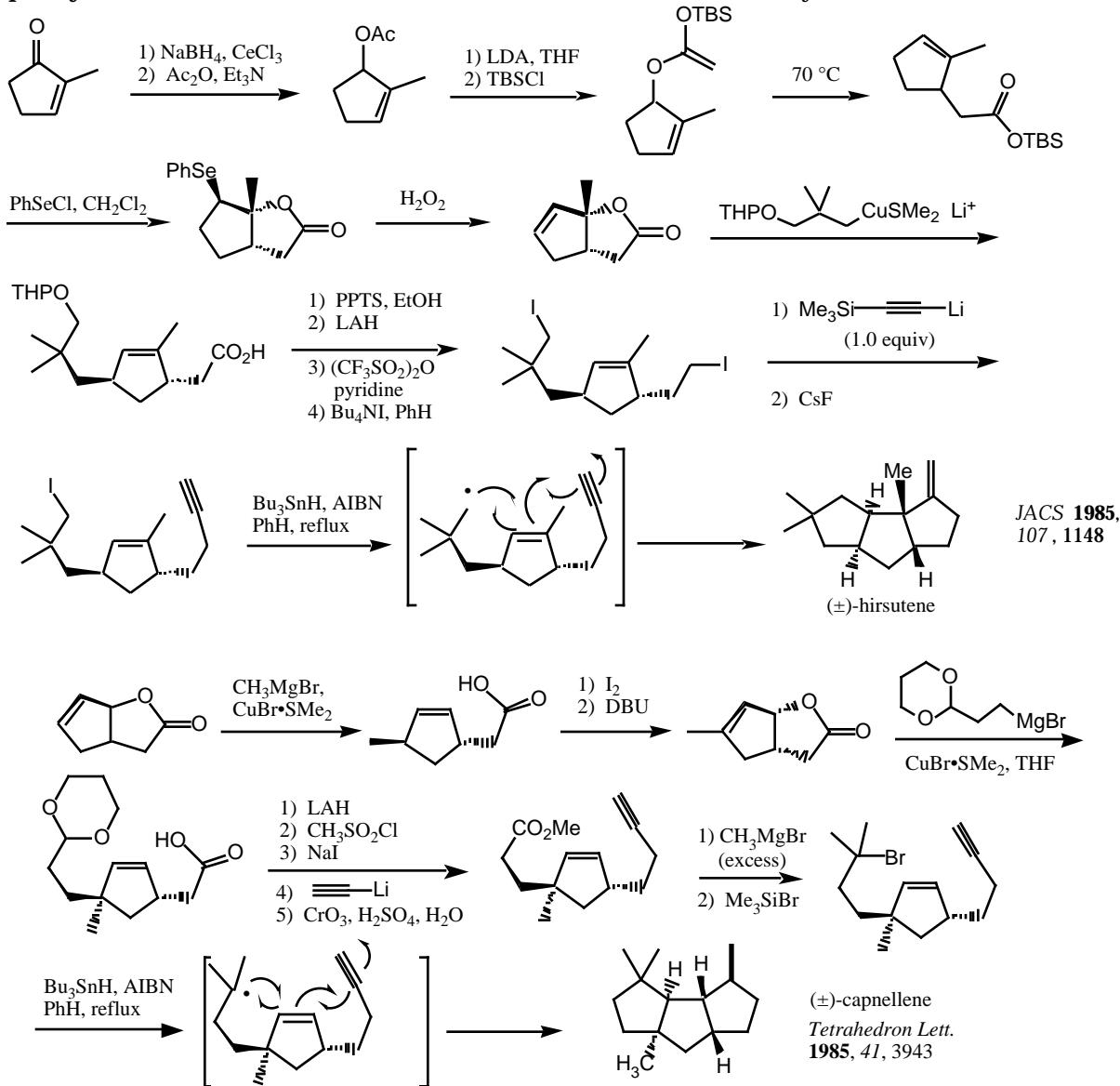
1-, or 3-substituted 5-hexenyl radicals give cis disubstituted cyclopentanes

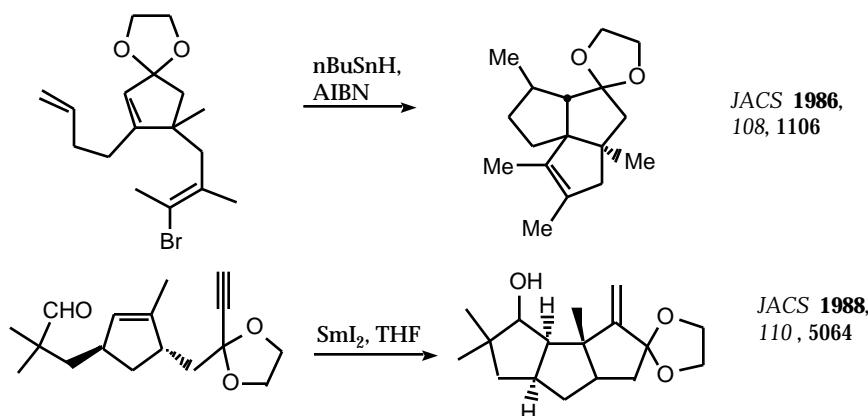
2-, or 4-substituted 5-hexenyl radicals give trans disubstituted cyclopentanes



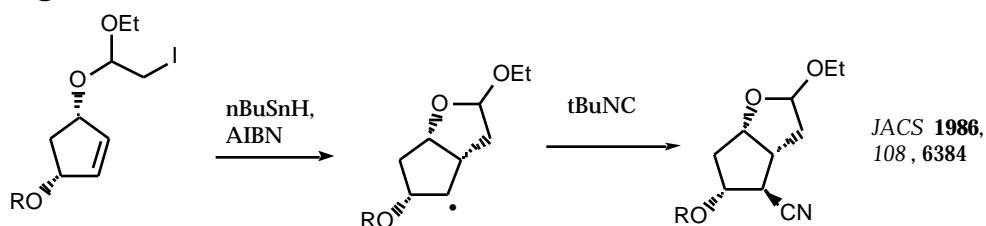


**multiple cyclizations:** D. Curran *Advances in Free Radical Chemistry* **1990**, 1, 121.

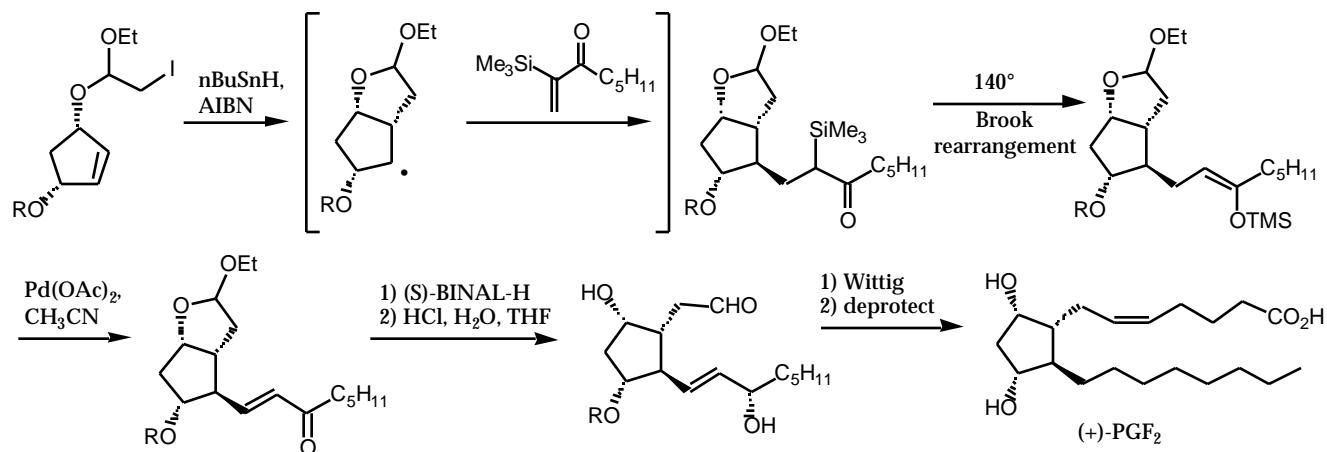




### radical trapping

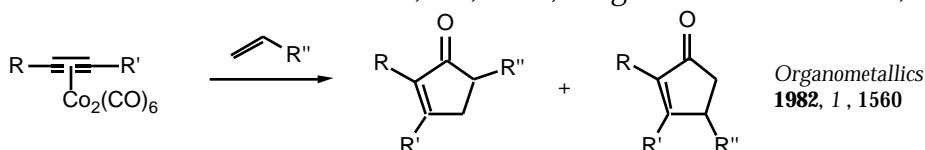


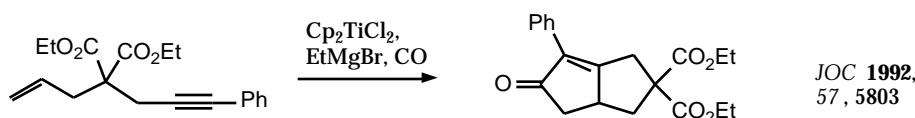
can also be trapped with acrylate esters or acrylonitrile.



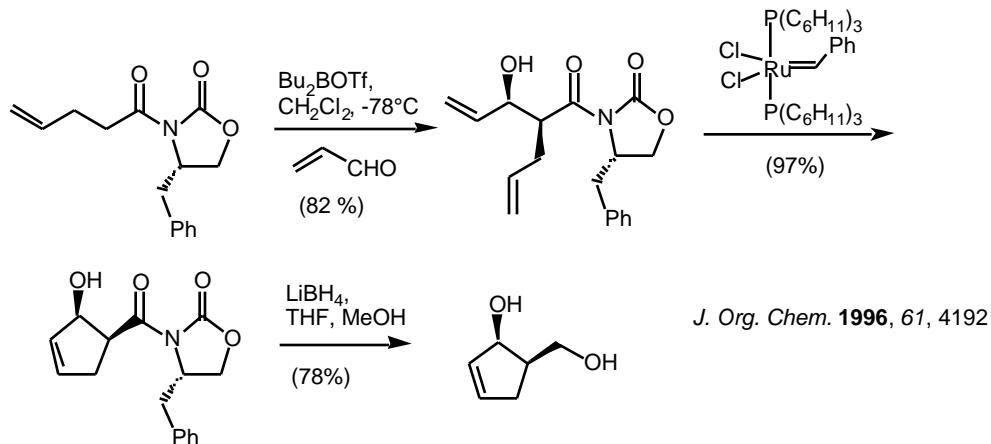
Paulson-Khand Reaction

Tetrahedron 1985, 41, 5855; Organic Reactions 1991, 40, 1.



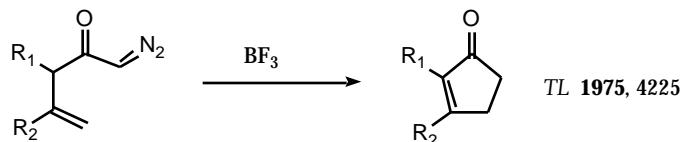


*Ring-Closing Metathesis* *Tetrahedron* 1998, 54, 4413, *Acc. Chem. Res.* 1995, 25, 446.



Diazoketones

*Tetrahedron* 1981, 37, 2407; *Organic Reactions* 1979, 26, 361



FVP of Acetylenic Ketones

